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LACTAM ACETALS : PART XXIV
REACTION WITH ACTIVATED HALOALKYL COMPOUNDS WITH AND
WITHOUT ZINC ¹

Sanjay Jain, Rahul Jain, Jujhar Singh and Nitya Anand *

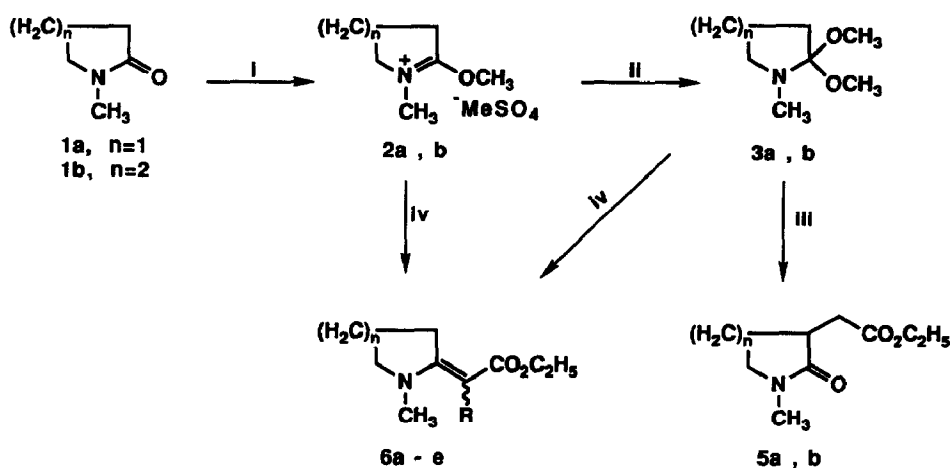
Medicinal Chemistry Division
Central Drug Research Institute
Lucknow 226001, India

Abstract: Reaction of 2-alkoxyimmonium methosulfates (2), and of lactam acetals (3) derived therefrom, with α -haloesters in presence of zinc (Reformatsky condition) yielded N-alkyl-2-(α -alkyl- α -alkoxy-carbonyl)methylene-1-azacycloalkanes (6), while reaction of 3 with α -haloesters without zinc gave 3-alkoxycarbonylmethyl-1-azacycloalkane-2-one (5). Similar reaction of 2 and 3 with 4-bromomethylquinolin-2-one (4) in presence of zinc gave N-alkyl-2-[4-(2-oxoquinolyl)methylene]-1-azacycloalkanes (7), a key intermediate for the synthesis of antimalarial quinoline-4-methanols.

Pyrrolidinyll and piperidinyll-acetic acids are important synthons commonly needed in organic synthesis²⁻⁴. If a carboxyalkyl residue could be conveniently grafted onto a pyrrolidinyll or piperidinyll substrate by a short path synthesis, these intermediates could become readily available. Pyrrolidone and piperidone acetals⁵ (or their immonium intermediates) can react as electrophile at C-2 and as nucleophile at C-3. While α -haloesters similarly act as C- α -electrophiles when used as such or as C- α -nucleophile under Reformatsky conditions⁶. The condensation of pyrrolidone and piperidone acetals (or immonium intermediates thereof) seemed to offer the possibility of direct obtention of the corresponding 2- or 3-carboxyalkyl derivatives depending upon the reaction conditions used. Similarly reaction of 4-bromomethylquinolin-2-one with lactam acetals under conditions where side chain methyl would be electrophilic (as under Reformatsky conditions) seemed to offer a convenient route to the synthesis of 4-[2-(1-azacycloalkyl)methyl]quinolinones, which are useful intermediates for synthesis of quinolinemethanol antimalarials⁷ related to quinine and mefloquine. The successful execution of these reactions is reported in this communication.

Condensation of ethyl bromoacetate [35 mmol] with lactam acetals 3a and b [20 mmol] in a sealed tube at 120° gave 3-carbethoxymethyl-1-

methyl-2-pyrrolidone/piperidone (**5a and b**)⁸ in excellent yields (Scheme 1, Table 1). On the other hand, when lactam acetals **3a and b** [12 mmol] were reacted with ethyl 2-bromoacetate/propionate/butyrate [10 mmol] in presence of freshly activated zinc dust [12 mmol] in Reformatsky reaction condition, the corresponding 2-(α -alkyl- α -ethoxycarbonyl)methylene-1-methylpyrrolidines/piperidines (**6a-e**)⁹ were obtained in 27-38% yield (Scheme 1, Table 1). With a view to further reduce the number of steps, the Reformatsky reaction was carried out on methoxyimmonium methosulfates **2a and b**, which are intermediates in the preparation of lactam acetals, when 2-(α -alkyl- α -ethoxycarbonyl)methylene-1-methylpyrrolidines/piperidines (**6a-d**) were obtained even more conveniently and in better yield (46-58%) (Scheme 1, Table 1). **6a and d** have been synthesized earlier by reacting (methylthio)alkylideniminium salt with ethyl acetate² or with Meldrum's acid³ whereas **6b** were prepared by alkylation⁴ of **6a**.



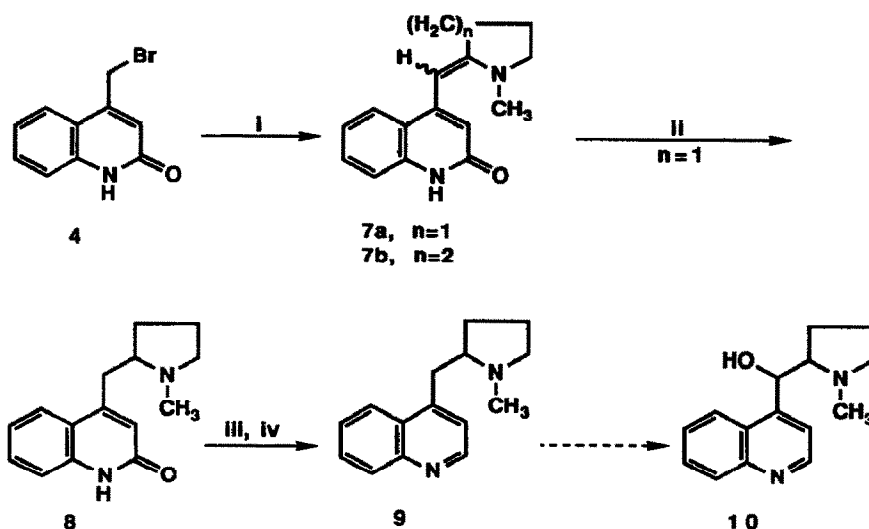
Reagents: I. Me_2SO_4 , 80° ; II. NaOMe/MeOH , 50° ; III. $\text{BrCH}_2\text{CO}_2\text{Et}$, 120° ;
 IV. Zn , $\text{BrCH(R)CO}_2\text{Et}$, H^+

SCHEME - 1

The reaction of 4-bromomethyl-1,2-dihydroquinolin-2-one¹⁰ [**4**, 30 mmol] with **2a and b** and/or **3a and b** [36 mmol] and activated zinc dust [39 mmol] in anhydrous THF (50 ml) was next carried out when 4α -(1-methyl-2-pyrrolidinylidene/piperidinylidene)methyl-1,2-dihydroquinolin-2-one (**7a and b**)¹¹ were obtained (Scheme 2, Table 1). Compound **7a** on catalytic hydrogenation (Pd-C catalyst at 3.5 kg/cm^3 H_2 pressure in a Parr hydrogenator) afforded the 4α -(1-methyl-2-pyrrolidinyl)methyl-1,2-dihydroquinolin-2-one (**8**)¹² in excellent yield. **8** on POCl_3 treatment followed by Pd-C hydrogenation in presence of triethylamine at RTP gave 4α -(1-

methyl-2-pyrrolidinyl)methylquinoline (9)¹³ in 77% yield. This easy availability of 4 α -(1-methyl-2-pyrrolidinyl)methylquinoline (9) which has earlier been oxygenated to quinolinemethanols 10⁷, provides a short and convenient approach to the synthesis of 4 α -heterocycloalkyl quinoline-4-methanols.

The Reformatsky reaction of α -haloesters with lactim ethers and lactim thioethers was also studied but proved unsuccessful.



Reagents: I. **2** or **3**, Zn, 80°, +H⁺; II. H₂, Pd-C, 45psi; III. POCl₃, 80°; IV. H₂, Pd-C, Et₃N, RTP

SCHEME - 2

Table 1 : Physical constants of compounds 5-7 #

Entry No. [^]	n	R	% yield	
			From 2	From 3
5a	1	-	-	89.5
5b	2	-	-	82.1
6a	1	H	58.3	35.5
6b	1	Me	51.7	33.9
6c	1	Et	46.2	27.9
6d	2	H	49.5	36.1
6e	2	Me	46.1	37.3
7a	1	-	63.0	56.5
7b	2	-	42.8	25.6

Products were purified by chromatography using Silica gel (230-400 mesh), and characterised by their IR, NMR, MS data and gave satisfactory elemental analysis.

[^]All compounds are oil in nature except 7a (mp. 254°)

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REFERENCES & NOTES

1. CDRI Communication No.5237.
2. Gugelchuk, M.M.; Hart, D.J.; Tsai, Y.-M.; *J. Org. Chem.* (1981) **46** 3671.
3. Celerier, J.P.; Richaud, M.G.; Lhommet, G.; *Synthesis* (1983) 195.
4. Celerier, J.P.; Marchalant, E.D.; Lhommet, G.; *J.Het. Chem.* (1984) **21**, 1633.
5. Anand, N.; Singh, J.; *Tetrahedron* (1988) **44**, 5975.
6. Rathke, M.W.; *Organic Reactions* (1975) **22**, 423.
7. Ohnmacht, C.J.; Patel, A.R.; Lutz, R.E.; *J. Med. Chem.* (1971) **14**, 926.
8. Spectral data of **5a**:: IR[Neat, cm^{-1}]: 1740 (ester), 1690 (carbonyl). ^1H NMR[CDCl_3 , δ ppm]: 1.25 (t, 3H, $J=6$ Hz, $-\text{CCH}_3$), 2.00-2.50 (m, 4H, $4-\text{CH}_2$ & $-\text{CHCH}_2$), 2.80-3.10 (m, 4H, $-\text{NCH}_3$ & $3-\text{CH}$), 3.30-3.60 (m, 2H, $5-\text{CH}_2$), 4.15 (q, 2H, $J=6$ Hz, $-\text{OCH}_2$). MS: m/z 185 (M^+). Analysis calcd for $\text{C}_9\text{H}_{15}\text{NO}_3$: C, 58.35; H, 8.16; N, 7.56. Found: C, 58.79; H, 8.03; N, 7.71%.
9. Spectral data of **6a**:: IR[Neat, cm^{-1}]: 1620 (carbonyl). ^1H NMR[CDCl_3 , δ ppm]: 1.20 (t, 3H, $J=6$ Hz, $-\text{CCH}_3$), 1.70-2.20 (m, 2H, $4-\text{CH}_2$), 2.70 (s, 3H, $-\text{NCH}_3$), 2.90-3.65 (m, 4H, $3,5-\text{CH}_2$), 4.05 (q, 2H, $J=6$ Hz, $-\text{OCH}_2$), 4.45 (s, 1H, olefinic-H). MS: m/z 169 (M^+). Analysis calcd for $\text{C}_9\text{H}_{15}\text{NO}_2$: C, 63.87; H, 8.93; N, 8.27. Found: C, 63.90; H, 8.87; N, 8.53%.
10. Cook, D.J.; Bowen, R.E.; Sorter, P.; Daniels, E.; *J. Org. Chem.* (1961) **26**, 4949.
11. Spectral data of **7a**:: IR[KBr, cm^{-1}]: 1625 (carbonyl). ^1H NMR[CDCl_3 , δ ppm]: 1.70-2.20 (m, 4H, $3'$ & $4'-\text{CH}_2$), 2.90 (s, 3H, $-\text{NCH}_3$), 3.20-3.50 (m, 2H, $5'-\text{CH}_2$), 5.22 (s, 1H, $=\text{CH}$), 6.32 (s, 1H, $3-\text{ArH}$), 7.00-7.90 (m, 4H, ArH). MS: m/z 240 (M^+). Analysis calcd for $\text{C}_{15}\text{H}_{16}\text{N}_2\text{O}$: C, 75.00; H, 6.66; N, 11.66. Found: C, 74.77; H, 6.33; N, 11.51%.
12. Spectral data of **8**:: yield 72.0%, IR[KBr, cm^{-1}]: 1620 (carbonyl). ^1H NMR[CDCl_3 , δ ppm]: 1.51-2.00 (m, 4H, $3'$ & $4'-\text{CH}_2$), 2.19-2.91 (m, 5H, $-\text{NCH}_3$ & $1'-\text{CH}_2$), 3.02-3.61 (m, 3H, $5'-\text{CH}_2$ & $2'-\text{CH}$), 6.68 (s, 1H, $3-\text{ArH}$), 7.00-7.92 (m, 4H, ArH). MS: m/z 242 (M^+). Analysis calcd for $\text{C}_{15}\text{H}_{19}\text{N}_2\text{O}$: C, 74.38; H, 7.43; N, 11.57. Found: C, 74.11; H, 7.07; N, 11.17%.
13. Spectral data of **9**:: yield 77.0%, IR[Neat, cm^{-1}]: 1600 (aromatic). ^1H NMR[CDCl_3 , δ ppm]: 1.60-2.10 (m, 4H, $3'$ & $4'-\text{CH}_2$), 2.60 (s, 3H, $-\text{NCH}_3$), 2.80-3.10 (m, 5H, $1'$ & $5'-\text{CH}_2$ & $2'-\text{CH}$), 7.28-7.35 (m, 1H, $3-\text{ArH}$), 7.58-8.20 (m, 4H, ArH), 8.80-8.90 (m, 1H, $2-\text{ArH}$). MS: m/z 226 (M^+). Analysis calcd for $\text{C}_{15}\text{H}_{18}\text{N}_2$: C, 79.64; H, 7.96; N, 12.38. Found: C, 79.42; H, 7.56; N, 12.00%.

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